





Synthetic Studies on Sarcodictyins and Eleutherobin: Synthesis of Fully Functionalized Cyclization Precursors

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Abstract: Unprecedented synthetic transformations were demonstrated during the preparation of fully functionalized cyclization precursors of type 2, in a synthetic approach to sarcodictyin A and B (1a,b) and eleutherobin (1c). © 1998 Elsevier Science Ltd. All rights reserved.

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The marine diterpenoids sarcodictyins A and B (1a,b) [1] and eleutherobin (1c) [2] (Scheme 1), extracted in minute quantities from the soft corals Sarcodictyon roseum and Eleutherobia albiflora, have shown outstanding biological activity. Both eleutherobin and sarcodictyins show potent in vitro cytotoxicity against diverse tumor cell lines [2b, 3] and compete with paclitaxel for its binding at the microtubuli, inhibiting their depolymerisation [3b, 4]. This tubulin stabilizing activity adds sarcodictyins and eleutherobin to the restricted family of taxol®-like cytotoxic agents (together with the epothilones [5] and discodermolide [6]).

Scheme 1

The extremely interesting biological activity and limited availability of sarcodictyins and eleutherobin from natural sources has elicited a rapid solution to the problem of their synthesis. Two total syntheses of eleutherobin have been accomplished recently [7], and Nicolaou et al. also synthesized sarcodictyin A and B [3b, 8], making use of a known elaboration of (+)-carvone to assemble the six-membered ring [9].

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On approaching the synthesis of the common tricyclic skeleton of sarcodictyins and eleutherobin via a C_2 - C_3 disconnection, intermediates of type 2a and 2b were identified as fully functionalized precursors to the closure of the B ring via an aldol type reaction (2a) or a McMurry reaction (2b). Their synthesis was achieved via a brief and convergent route, making use of some new synthetic transformations.

Intermediate 6 (Scheme 2), possessing the correct stereochemistry and appropriate functionality for the synthesis, was accessed through an unprecedented route, starting from the readily available natural product (-)-carvone. (-)-Carvone was subjected to selective hydrogenation to afford 3 [10]. Electrophilic formylation occurred from the less hindered face of the intermediate dienoxysilane [11], yielding 4 as the only diastereoisomer. Diastereoselective reduction from the equatorial direction with L-selectride[®] [12] gave alcohol 5 as the main product (9:1 diastereoisomeric ratio),² which was subjected to acetylation followed by an Ireland-Claisen rearrangement.

Scheme 2. Reagents and Conditions: a) MeMgBr (3M in Et₂O), FeCl₃, Et₂O, TMSCl, TEA, DMPU, -20 °C, RT, 81%; b) BF₃·Et₂O, (MeO)₃CH, CH₂Cl₂, -70 °C, 45%; c) L-selectride[®], THF, -78 °C, 91%, 9:1 d.r.; d) Ac₂O, TEA, DMAP, CH₂Cl₂, 95%; e) i. LiN(TMS)₂, TBDMSCl, THF, DMPU, -78 °C, RT; ii. Xylene, 190 °C, 5h; iii. NaOH, THF, 70%.

The acid appendage was elongated through either a *trans*- or a *cis*-selective Horner-Wittig reaction [13] with propionate derivatives on aldehyde 7 (Scheme 3). Following unsuccessful attempts with asymmetric dihydroxylation,³ oxidation of the double bond was obtained *via* Sharpless asymmetric epoxidation (SAE) of

Scheme 3. Reagents and Conditions: a) LiAlH₄, Et₂O, RT, 91%; b) Dess-Martin periodinane, CH₂Cl₂, RT, 100%; c) (CF₃CH₂O)₂P(O)CH(CH₃)CO₂Et, 18-crown-6·CH₃CN, KN(TMS)₂, THF, -78 °C, 85%, Z only; d) LiAlH₄, Et₂O, RT, 96%; e) i. (EtO)₂P(O)CH(CH₃)CO₂Et, NaH, DME, RT, 85%, 9:1 E/Z; ii. separation of major E isomer by flash-chromatography; f) (+)- or (-)-DET (12%), Ti(OiPr)₄ (10%), tBuOOH (5.5 M in nonane) (2 eq), CH₃Cl₃, -20 °C, 50-95%, see text for diastereoselectivities.

² The complementary ratio (1:9) was obtained by reduction from the axial direction with LiAlH₄.

³ Although p-methoxybenzoyl protected allylic alcohols are reported to undergo regionselective dihydroxylation [14], the p-methoxybenzoyl derivative of 8 furnished the product corresponding to dihydroxylation of the endocyclic double bond when subjected to standard AD conditions.

allylic alcohols 8 and 9, prepared *via* reduction of the corresponding esters. Both (+)- and (-)-diethyl tartrate (DET) were employed, so that all possible diastereoisomeric epoxy-alcohols (10a,b, 11a,b) were accessed. Reaction of 8 with (+)-DET gave a 1:9 mixture of 10a and 10b, while the reaction with (-)-DET proved to be a mismatched case, yielding a 1:1 ratio of the two diastereoisomers. On the contrary, excellent stereocontrol was obtained for 9, which afforded either 11a or 11b upon reaction in the presence of (-)- and (+)-DET respectively (≥ 98:2 diastereoisomeric ratios). All epoxy-alcohols were cleanly oxidized to the corresponding aldehydes (e.g. 12a,b, Scheme 4) with buffered Dess-Martin periodinane.

Although the anion of ethyl 4-(triphenylphosphoranylidene)acetacetate (13) is known to perform a Wittig reaction with aldehydes [15], no reaction was obtained with 12a,b under the reported conditions (NaH, THF, H₂O). However, when 12a was exposed to the anion of 13, preformed with sodium bis-trimethylsilylamide in THF, a rapid and Z-selective Wittig reaction ensued; the intermediate enolate performed a conformationally biased intramolecular attack on the epoxide ring at the more substituted carbon, and 14a was isolated in 60% overall yeld. This unprecedented transformation allowed the introduction of all the remaining carbon atoms and of the dihydrofuran ring C of the target molecules in a single highly convergent step. Alternatively, since 10a represents a mismatched case product of the SAE, we examined 11a as an alternative: the Wittig reaction of 12b with 13 affords 14b, from which the correct stereochemistry can be accessed via simple inversion of the secondary alcohol. Both alcohols were easily protected with triethylsilyl chloride.

Scheme 4. Reagents and Conditions: a) Dess-Martin periodinane, Py, CH₂Cl₂, RT, 98%; b) 13, NaN(TMS)₂, THF, 0 °C, RT, 70%, 85:15 Z/E; c) TESCl, Imidazole, CH₂Cl₂, RT, 75%.

In view of employing an intramolecular McMurry reaction for the formation of ring B, we examined elaboration of the dihydrofuran side chain on model compound 16 (Scheme 5), which was obtained via an analogous route from hydrocinnamic aldehyde. Reduction of the ester moiety with lithium aluminium hydride afforded allylic alcohol 17, an electron rich enol ether. It is known that the directive effect of hydroxy groups in dimethyldioxirane (DMDO) oxidations is diminished in protic solvent mixtures [16]. However, by treating 17 with DMDO in MeOH/acetone, exclusive oxidation of the exocyclic double bond was obtained, and the intermediate epoxide was smoothly converted to the desired diol 18 (mixture of diastereoisomers) by regioselective solvolysis in 92% overall yield. Ketone 20a (1.8:1 acetal epimeric ratio) was easily obtained by selective protection of the primary hydroxy group and oxidation with Dess-Martin periodinane.

Scheme 5. Reagents and Conditions: a) LiAlH₄, Et₂O, 0 °C; b) DMDO, Acetone, MeOH, 0 °C, 92% (overall yield for a and b); c) TBDPSCl, Imidazole, CH₂Cl₂, RT, 80%; d) Dess-Martin periodinane, Py, CH₂Cl₂, RT, 83%.

The cyclization reactions of compounds 15b and 20b are currently under study, as well as new synthetic approaches to the tricyclic skeleton of sarcodictyins and eleutherobin.

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